

## REVIEW

# Mitigation Option of Greenhouse Gas Emissions from Livestock Manure Composting

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### Abstract

Composting of livestock manure is a significant source of greenhouse gases such as nitrous oxide (N<sub>2</sub>O), but the complexity of the N<sub>2</sub>O generation pathway hinders efforts to develop effective countermeasures. In this paper, we present one mitigation option for N<sub>2</sub>O emissions from composting. Nitrite (NO<sub>2</sub><sup>-</sup>) has an important role in N<sub>2</sub>O generation. It has been observed that significant N<sub>2</sub>O emissions from composting are induced when NO<sub>2</sub><sup>-</sup> is accumulated during nitrification. Accordingly, in our technique, nitrite-oxidizing bacteria (NOB), as nitrifying bacteria, are added in the middle of composting fermentation to prevent NO<sub>2</sub><sup>-</sup> accumulation. Adding NOB prevents prolonged NO<sub>2</sub><sup>-</sup> accumulation, which had resulted in low N<sub>2</sub>O emissions. This technique is also cost-effective, because mature compost can be used as an NOB source. Mitigating N<sub>2</sub>O emissions allows more nitrate nitrogen (NO<sub>3</sub><sup>-</sup>), of high value as fertilizer, to be preserved in the compost product. Moreover, it was confirmed that this technique could be combined with that for NH<sub>3</sub> mitigation using chemical reagents. Before actual use, it is necessary to investigate the methods used to determine optimal timing to add NOB source and prevent N<sub>2</sub>O from being generated from an NO<sub>3</sub><sup>-</sup> reduction.

**Discipline:** Agricultural environment, animal industry

**Additional key words:** nitrification, nitrite-oxidizing bacteria, nitrous oxide

### Introduction

Composting is a principal means of treating organic waste such as livestock manure. However, during the composting process, substantial amounts of harmful environmental gases are emitted, including greenhouse gases such as nitrous oxide (N<sub>2</sub>O) (Czepiel et al. 1996, Fukumoto et al. 2003a, Osada et al. 2000, Sommer 2001). Accordingly, reducing the emission of harmful gases from the composting process has become increasingly important.

N<sub>2</sub>O is a powerful greenhouse gas, with a global warming effect around 300-fold stronger per molecule than carbon dioxide (IPCC 2001). Moreover, N<sub>2</sub>O also impacts on ozone layer depletion (Crutzen 1981). Agriculture is the largest source of anthropogenic N<sub>2</sub>O emissions, of which livestock activity in particular makes a significant contribution (FAO 2006). Accordingly, it is important to develop a

technique to reduce N<sub>2</sub>O emissions from livestock activity, including composting of livestock manure. However, there are relatively few countermeasures to reduce N<sub>2</sub>O emissions from composting compared to NH<sub>3</sub> emissions.

The authors have been trying to develop countermeasures to reduce N<sub>2</sub>O emissions from composting of livestock manure. In this paper, we present the mitigation option and discuss issues concerning its actual usage.

### N<sub>2</sub>O generation in the composting process

N<sub>2</sub>O is generated via both nitrification and denitrification processes as intermediate products or by-products during the composting process. The lack of nitrite/nitrate contained in fresh manure means the nitrification start is a prerequisite for N<sub>2</sub>O generation from the composting process. Nitrification is performed by two kinds of bacteria, i.e.

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Received 23 October 2014; accepted 22 January 2015.

ammonia-oxidizing and nitrite-oxidizing (AOB and NOB). These bacteria cannot be active when exposed to high temperature and high free ammonia; their growth begins after the high-temperature period of composting fermentation.

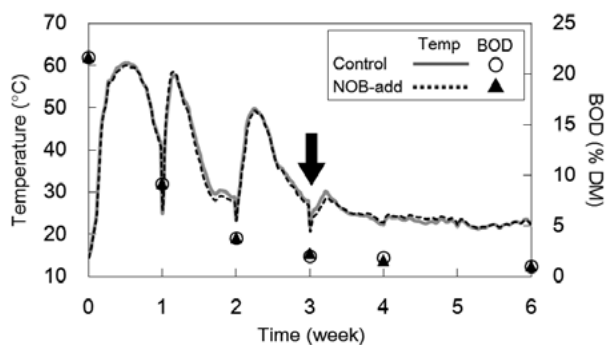
During the composting process, nitrite ( $\text{NO}_2^-$ ) is key to  $\text{N}_2\text{O}$  generation (He et al. 2001). Accordingly, there is potential to mitigate  $\text{N}_2\text{O}$  emissions by controlling  $\text{NO}_2^-$ . Based on this hypothesis, we conducted several composting experiments.

### Mitigating $\text{N}_2\text{O}$ emissions by adding nitrite-oxidizing bacteria

To mitigate  $\text{N}_2\text{O}$  emissions by controlling  $\text{NO}_2^-$ , we attempted the complementary addition of NOB of nitrifiers midway through the composting process. In our experiments, it was confirmed that the prolonged accumulation of  $\text{NO}_2^-$  tended to emerge during swine manure composting, underlining the suitability of the experiment to determine the effect of NOB addition on  $\text{NO}_2^-$  accumulation and  $\text{N}_2\text{O}$  emissions.

Our first study on the effect of NOB addition in swine manure composting was conducted using a laboratory-scale apparatus (Fukumoto et al. 2006). Continuous ventilation was conducted at a fixed rate (10 L/min) and the  $\text{N}_2\text{O}$  concentration in the exhaust air from the apparatus was continuously measured by an infrared photoacoustic detector (IPD, multigas monitor type 1312, Innova, Denmark). In this experiment, we used mature swine compost as the NOB source added after the thermophilic composting phase (Fig. 1), which contained around  $10^6$  NOB cells per gram.

Changes in the number of nitrifying bacteria (AOB and NOB) were shown in Figure 2. During the thermophilic composting phase, there were few nitrifying bacteria, but AOB subsequently began to increase. However, the growth of indigenous NOB was significantly delayed compared to



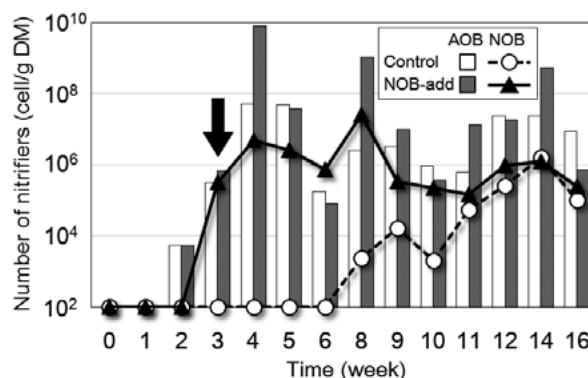
**Fig. 1. Changes in the temperature of composting material and BOD concentration during swine manure composting.**

Closed arrow indicates the addition of an NOB source of mature swine compost.

AOB in the control. Conversely, the cell density of the NOB population remained high after adding the NOB source.

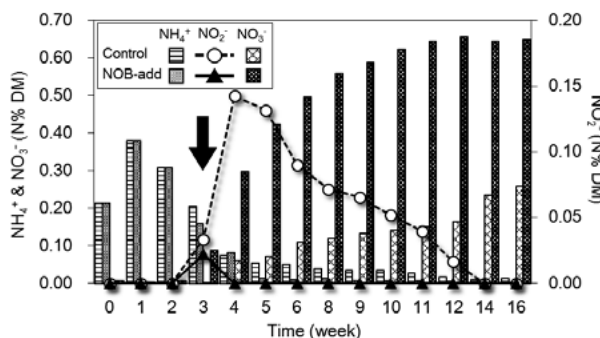
Changes in concentrations of inorganic nitrogen compounds were shown in Figure 3. After the AOB growth,  $\text{NH}_4^+$  oxidation started, but because of NOB absence,  $\text{NO}_2^-$  oxidation stalled, resulting in  $\text{NO}_2^-$  accumulation in the control. Conversely, with NOB addition,  $\text{NO}_2^-$  was smoothly oxidized into  $\text{NO}_3^-$  and no prolonged  $\text{NO}_2^-$  accumulation was observed. Moreover, it emerged that artificially adding the NOB source could prevent prolonged  $\text{NO}_2^-$  accumulation during swine manure composting.

In this experiment, we added the mature swine compost at an additional rate of 10% (w/w), which rapidly led to complete nitrification. However, a similar result was also obtained in another experiment with a lower additional rate (2% w/w) of mature swine compost, which contained around  $10^5$  NOB cell per gram (Fukumoto et al. 2011). Blouin et al. (1990) showed that an NOB concentration of at least  $10^5$  cell/ml was required before the complete oxidation



**Fig. 2. Changes in the number of nitrifiers during swine manure composting.**

Closed arrow indicates the addition of an NOB source of mature swine compost.



**Fig. 3. Changes in the concentrations of inorganic nitrogen compounds during swine manure composting.**

Closed arrow indicates the addition of an NOB source of mature swine compost.

of  $\text{NO}_2^-$  during the incubation of swine waste. However, because the population size of NOB can be grown even after the addition, we do not consider it crucial to fill the required cell number for complete nitrification when adding the mature compost. Adding mature compost during composting fermentation affects the workload, so further study is necessary to determine a suitable rate at which to add mature compost as the NOB source.

Emission patterns of  $\text{N}_2\text{O}$  were shown in Figure 4.  $\text{N}_2\text{O}$  emissions started after the thermophilic composting phase. In the control,  $\text{N}_2\text{O}$  emissions continued for an extended period, during which  $\text{NO}_2^-$  remained in the compost material. In contrast,  $\text{N}_2\text{O}$  emissions ceased within 1 week of adding the NOB source. Consequently, the total  $\text{N}_2\text{O}$  emissions during the composting experiment declined by 80% in the case of NOB addition compared to the control. Accordingly, we considered that a quantitative reduction effect of  $\text{N}_2\text{O}$  emissions from swine manure composting could be expected by adding the NOB source.

#### Effect of mitigating $\text{N}_2\text{O}$ emissions on nitrogen preservation

Nitrogen loss during composting not only induces environmental problems but also reduces the compost value as a fertilizer. The key cause of nitrogen loss in composting is thought to be  $\text{NH}_3$  emissions during the thermophilic phase. However,  $\text{NH}_3$  volatilization and denitrification were reportedly of similar magnitude when composting swine manure (Petersen et al. 1998). In particular, as already shown in this paper, the quantity of nitrogen loss may vary significantly, with or without  $\text{NO}_2^-$  accumulation. Accordingly, we examined the nitrogen mass balance during swine manure composting with the treatment of adding NOB source (Fukumoto & Inubushi 2009).

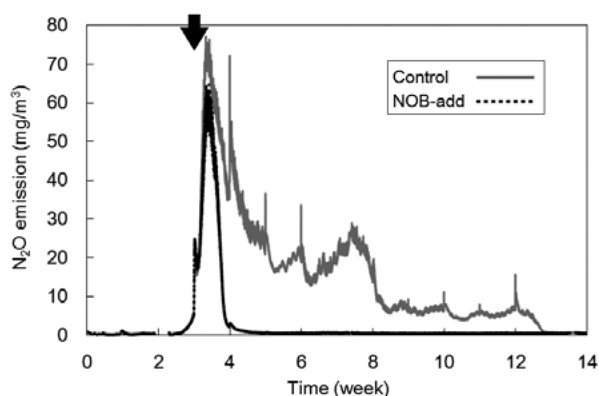


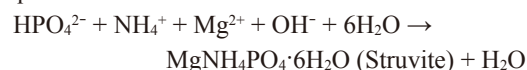
Fig. 4. Emission patterns of  $\text{N}_2\text{O}$  during swine manure composting.

Closed arrow indicates the addition of an NOB source of mature swine compost.

During the  $\text{NO}_2^-$  accumulation in swine manure composting, the increasing curve of  $\text{NO}_3^-$  was gentle,  $\text{NO}_3^-$  but increased quickly when the NOB source was added. Consequently, the  $\text{NO}_3^-$  content in the final product exceeded the control when adding the NOB source (Fig. 3). Moreover,  $\text{NO}_2^-$  accumulation induced not only  $\text{N}_2\text{O}$  emissions but also other nitrogenous emissions (Fig. 5). Accordingly, by avoiding  $\text{NO}_2^-$  accumulation, nitrogen loss during the maturation composting phase could be effectively reduced and  $\text{NO}_3^-$  nitrogen with high fertilizer value was preserved at a higher level in the final product.

#### Combination with struvite crystallization to reduce $\text{NH}_3$ emissions

There are several countermeasures to reduce  $\text{NH}_3$  emissions from composting fermentation (Burrows 2006, Kuroda et al. 2004, Yasuda et al. 2009). Struvite crystallization is one of the effective countermeasures for  $\text{NH}_3$  emissions. Struvite is crystallized magnesium ammonium phosphate (MAP), which is formed according to the following equation:



Adding magnesium (Mg) and phosphate ( $\text{PO}_4$ ) salts is an effective way to trigger struvite crystallization in the composting process (Jeong & Kim 2001). Combining with this technique allows nitrogen loss throughout the composting process to be mitigated. However, as had been reported, the reagent addition of struvite crystallization enhancement adversely impacts on microorganisms of composting fermentation (Jeong & Hwang 2005), while

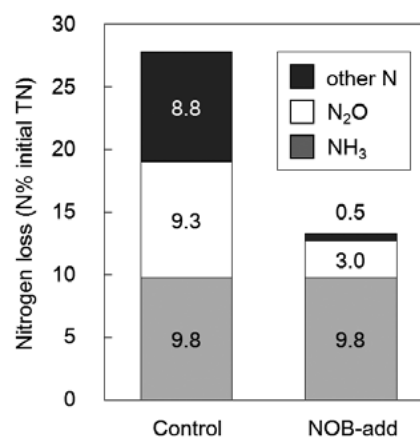


Fig. 5. Details of nitrogen losses during swine manure composting.

Nitrogen loss is indicated by the ratio of lost nitrogen based on total nitrogen (TN) in the initial compost material.

the technique of adding NOB to mitigate N<sub>2</sub>O emissions depends on microbial activity. Accordingly, we conducted a composting experiment to investigate the potential for technical coexistence.

To quantify the combined effect of struvite crystallization and adding NOB source on the mitigation of nitrogenous emissions, swine manure composting experiments were conducted (Fukumoto et al. 2011). To enhance the struvite crystallization, MgCl<sub>2</sub>·6H<sub>2</sub>O and H<sub>3</sub>PO<sub>4</sub> were added at the start of composting, while mature swine compost was added as the NOB source after the thermophilic composting phase.

The NH<sub>3</sub> emissions declined when the reagents were added (Fig. 6). To confirm struvite crystallization, the amount of nitrogen fixed in struvite crystal was measured according to the procedure of Tanahashi et al. (2010). Consequently, the amount of nitrogen fixed in struvite crystal was shown to be higher when treating reagent addition (Fig. 7), which meant the struvite crystallization was considered to help reduce NH<sub>3</sub> emissions in this experiment.

Adding the NOB source reduced N<sub>2</sub>O emissions, even after adding reagents for struvite crystallization (Fig. 6), while adding reagents did not affect the activity of nitrifying bacteria. Accordingly, it was thought that the technique of adding an NOB source could coexist with struvite crystallization.

### Issues for practical usage

The effect of adding NOB source on the reduction in N<sub>2</sub>O emission from the composting process was investigated and a quantitative reduction effect on N<sub>2</sub>O emissions is expected using this technique (Fig. 8). However, due attention regarding the timing of NOB source addition is nec-

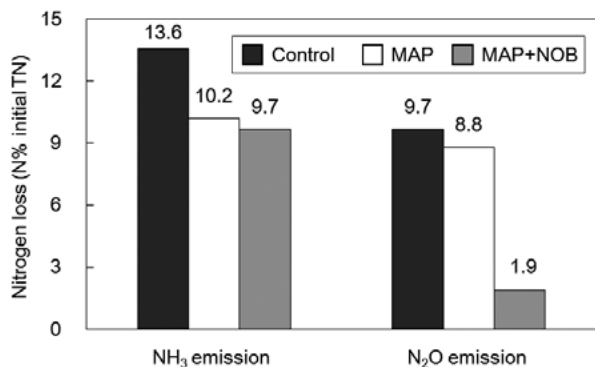


Fig. 6. Nitrogen losses by NH<sub>3</sub> and N<sub>2</sub>O emissions during swine manure composting.

Nitrogen loss is indicated by the ratio of lost nitrogen based on total nitrogen (TN) in the initial compost material.

essary. In our previous experiment, adding the NOB source during the thermophilic composting phase resulted in the extinction of NOB. Accordingly, this must be avoided and a method developed to determine the optimal time at which to add the NOB source during the composting process.

The effect of adding NOB source on mitigating N<sub>2</sub>O emissions was confirmed in small-scale composting experiments. Accordingly, a larger scale experiment was necessary before actual use. As shown in this paper, this technique reduces N<sub>2</sub>O emissions induced by NO<sub>2</sub><sup>-</sup> accumulation during the nitrification process. However, N<sub>2</sub>O generated from the NO<sub>2</sub><sup>-</sup>/NO<sub>3</sub><sup>-</sup> reduction process, i.e. denitrification cannot be mitigated by this technique. In particular, the increased scale of the compost pile causes the anaerobic portion inside the compost pile to expand (Fukumoto et al. 2003b). Accordingly, when this technique is applied in actual composting, it is important to minimize the N<sub>2</sub>O derived from the denitrification process. For example, it may be effective to reduce the frequency of unnecessary pile turning after adding the NOB source. Based on the above issues, it is important to verify the effect of this technique in actual-scale composting.

### Conclusion

During the composting process, prolonged NO<sub>2</sub><sup>-</sup> accumulation induces significant N<sub>2</sub>O emissions. The accumulation is caused by the delayed growth of indigenous NOB compared to AOB. Adding the NOB source (mature compost) in the middle of the composting process could prevent prolonged NO<sub>2</sub><sup>-</sup> accumulation, which reduced N<sub>2</sub>O emissions. This technique produces valuable compost with high nitrogen content and is expected to preserve greater amounts of nitrogen, alongside measures to reduce NH<sub>3</sub>

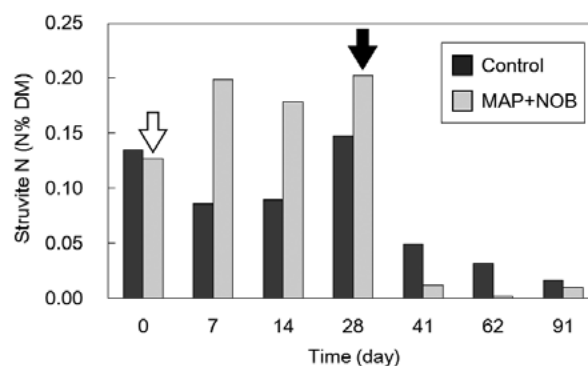


Fig. 7. Changes in the concentrations of nitrogen fixed in struvite crystal during swine manure composting.

Open arrow indicates the addition of reagents for struvite crystallization. Closed arrow indicates the addition of an NOB source of mature swine compost.

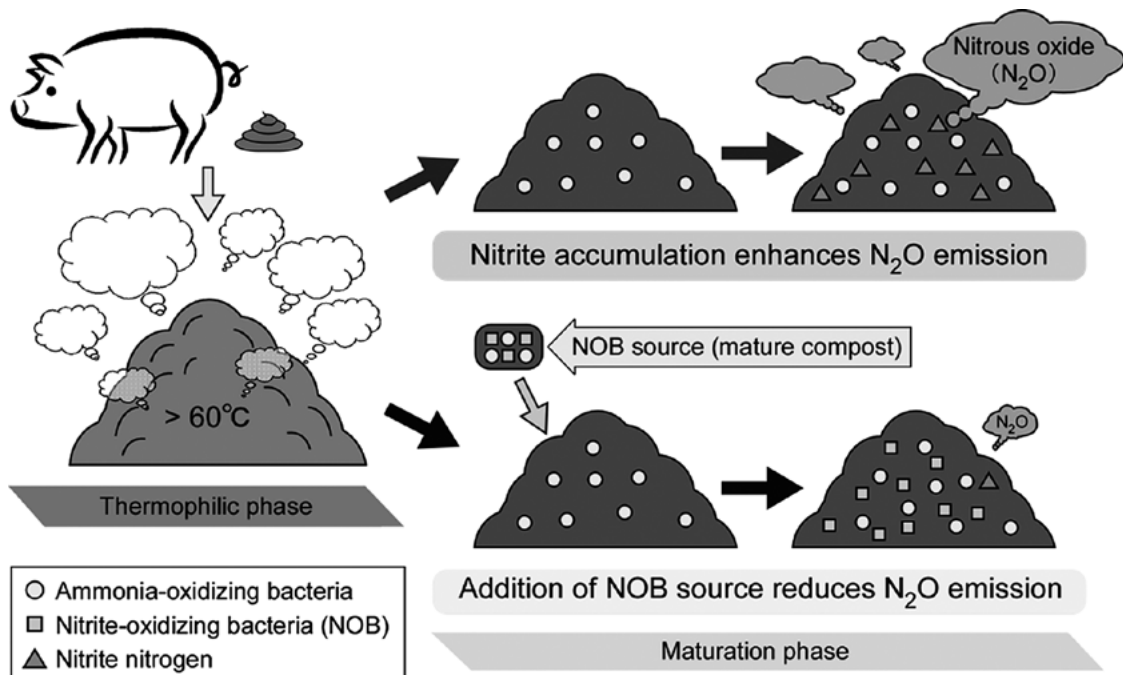


Fig. 8. Schematic of the bioremediational technique of mitigating N<sub>2</sub>O emissions during the composting process.

emissions. To apply this technique to actual-scale composting, it is necessary to investigate methods to determine suitable times at which to add the NOB source and prevent N<sub>2</sub>O from being generated from denitrification.

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